

A. Houbi^{1, 2*}, A.A. Zharmenov^{1, 2}, Y. Atassi³,
Z.T. Bagasharova^{1, 2}, S. Mirzalieva^{1, 2}, B.A. Karibayev⁴

¹National Center on Complex Processing of Mineral Raw Materials of the Republic of Kazakhstan, Almaty, Kazakhstan;

²Department of Chemical Technology of Inorganic Substances, Al-Farabi Kazakh National University, Almaty, Kazakhstan;

³Department of Applied Physics, Higher Institute for Applied Sciences and Technology, Damascus, Syria;

⁴Department of Physics and Technology, Al-Farabi Kazakh National University, Almaty, Kazakhstan

(*Corresponding author's e-mail: Anashoubi@gmail.com)

Synthesis and Microwave Absorption Properties of (Ni_{0.5}Zn_{0.5}Fe₂O₄/CI/CB) Ternary Composites

In this work, ternary composites of NiZn ferrite/carbonyl iron/carbon black (Ni_{0.5}Zn_{0.5}Fe₂O₄/CI/CB) were prepared via two stages. Firstly, Ni_{0.5}Zn_{0.5}Fe₂O₄ was prepared by a self-combustion method using sucrose as a fuel. After that, the operation was continued via mixing CB, CI, and Ni_{0.5}Zn_{0.5}Fe₂O₄ through grinding balls. Three various weight ratios of Ni_{0.5}Zn_{0.5}Fe₂O₄/CI/CB (1:1:1, 1:1:2, and 2:1:1) with various thicknesses (2–6 mm) were prepared. The absorbers were prepared by dispersing (Ni_{0.5}Zn_{0.5}Fe₂O₄/CI/CB) composites with a weight ratio within a paraffin wax matrix of 40 % w/w. X-ray diffractometry and FTIR spectroscopy were used in order to characterize the samples. The morphology of the powders was investigated by SEM. The functional characterization was accomplished by measuring the microwave absorption properties in the frequency band of 8.8–12 GHz. The microwave absorption materials (MAMs) showed wide bandwidths under –10 dB in the range of 2.81–3.20 GHz and reasonable surface density in the range of 3.625–4.041 kg/m². The absorber of 3.20 GHz bandwidth had a minimal reflection loss of –19.4 dB at the matching frequency of 9.92 GHz with a thickness of 6 mm.

Keywords: NiZn ferrite; Carbonyl iron, Carbon black, Reflection loss, Absorption bandwidth, Matching frequency, Paraffin wax matrix, Surface density.

Introduction

Electromagnetic interference (EMI) would be regarded as an unwanted result of modern technology that has dangerous effects on human health, intelligent devices, and military industries. Consequently, this EMI has become a critical worldwide issue and its alleviation could be accomplished only by utilize of EMI shielding materials. Nowadays, several magnetic loss materials such as hexagonal ferrites, spinel ferrites, and carbonyl iron or dielectric loss materials such as conductive polymers and carbonaceous materials have played a significant role in high-frequency EM wave absorption. Nevertheless, the drawbacks involving elevated density, low reflection absorption, and narrow wideband have hugely limited conventional loss materials' workable benefits for EM wave absorption. In recent years, MA composites based on carbon, carbonyl iron, and ferrite, have obtained significant concern due to their excellent electrical and ferrimagnetic characteristics. Carbonaceous materials-based composites have pulled in major attention for microwave absorption lately because of the unique structure of carbon-based materials. Carbon black is usually used to fit the requirements of high-effective microwave attenuation materials because of its superior characteristics, for example, high permittivity, high specified surface region, huge interface, etc. Carbon black has a unique place in the band of elevated-frequency MAMs. Furthermore, spinel ferrites and carbonyl iron have excellent MA characteristics due to their unique magnetic characteristics. NiZn ferrites and carbonyl iron are considered suitable materials for high-frequency implementations [1, 2]. When NiZn ferrite and carbonyl iron are mixed with carbon black, the MA characteristics of the resultant composite are anticipated to enhance. According to this, BaFe₁₂O₁₉/CI absorbers with various powder ratio compositions in the frequency range of 2–18 GHz were successfully prepared by Feng et al. [3]. The single-layer and double-layer absorbers were prepared, and their MA characteristics were studied. The outcomes showed that the double-layer absorber was clearly more than that of the single-layer absorber. Where the reflection loss (RL) for the double-layer absorber was –13 dB in the frequency range (6–18 GHz) and less than –8 dB in the frequency range (2–18 GHz). The thicknesses of the absorbers were 3.6 and 3.7 mm, respectively. On the other hand, Yan et al. [4] prepared a

mixture of doped polyaniline (PANI) coated porous structure carbonyl iron powder (CIP) and graphene sheets. The results showed that the absorption bandwidth under -10 dB ($BW_{-10\text{ dB}}$) was 4.6 GHz for 2 mm thickness for the composites with 40 wt.% of PANI@ porous CIP and 5 wt.% of Graphene. The outcomes showed that graphene-PANI@ porous CIP was a promising wave absorbing composite material. Until now, to the best of our knowledge, no studies have been reported on the MA properties of composites made up of NiZn ferrite/carbonyl iron/carbon black ($\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$). In this work, a perfect absorber was obtained by incorporating $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ and CI (magnetic loss materials) and CB (dielectric loss material) within a paraffin wax matrix. Where we study the effect of different weight ratios of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ and its effect on the MA properties. A distinct feature of this work is that NiZn ferrite was synthesized through a self-combustion method utilizing polyvinyl alcohol (PVA) as a chelating agent. After that, the operation is continuous through mixing and grinding CB, CI, and NiZn ferrite by grinding balls.

Experimental

Chemicals: polyvinyl alcohol (PVA) (hydrolyzed, MW: 72000), nickel(II) nitrate hexahydrate ($\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, 98.3 % purity), zinc nitrate hexahydrate ($\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, 98.7 % purity), and iron(III) nitrate nonahydrate ($\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$, 98.2 % purity) were purchased from TRADING COMPANY ANT, Russia. On the other hand, carbonyl iron (CI, 99.6 % purity) and carbon black (CB, 99.5 % purity) were purchased from Cabot Norit Company, Netherland.

Instruments used: A powder X-ray diffractometer (XRD, Rigaku Miniflex 600, Cu-K α) was used for determining the crystal structures of the powders. Fourier Transform IR (FTIR) spectra were recorded on a Perkin Elmer spectrum 65 FTIR spectrometer in the range of 400–4000 cm^{-1} . A scanning electron microscope (FEI Quanta 200 3D) was used for determining the powders morphology. The microwave absorption properties of the prepared samples were calculated by using the horn antenna connected to an oscilloscope (AKTAKOM ADS-2221M).

Methodology:

1. Synthesis of NiZn ferrite, carbonyl iron and carbon black powders

Ferrite ($\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$) nanoparticles were prepared by a self-combustion method. $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ were synthesized by taking appropriate amounts of ($\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$), ($\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$), and ($\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$). They were blended together with an aqueous solution of sucrose (2 moles per metal ion) and 1 % an aqueous solution of polyvinyl alcohol. The whole mixture was blended totally and heated at 90 °C for 7 h to shape a viscous liquid. The heating process was accompanied by the evolution of brown fumes of NO_2 from the decomposed metal nitrate salts. Then, the mixture was transferred to the furnace for drying for 2 h at 200 °C to obtain a fluffy carbonaceous pyrolyzed mass. After that, the resulting mass was annealed for 4 h at 650 °C to obtain nanoparticles of ferrite. Typical images of a prepared ferrite by a self-combustion method are shown in Figure 1. On the other hand, CI and CB were purchased from Cabot Corporation Company. The average particle size of CI and CB powders was measured utilizing the sieve shaker and it was between 10–25 μm and 2–8 μm , respectively. CI and CB powders were milled for 12 h at 300 rpm via grinding balls to obtain fine powders.

2. Synthesis of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ composites

Ferrite nanoparticles were mixed and milled with CI and CB powders by grinding balls. Three various weight ratios of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ (1:1:1, 1:1:2, and 2:1:1) were synthesized. The $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ composites were milled for 1 h at 300 rpm.

3. Preparation of absorber samples

Paraffin wax was symmetrically blended with $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ composites powders with a weight ratio of 40 % w/w within a paraffin wax matrix by heating and stirring for 15 min. Afterward, the single-layer samples were molded to the dimensions of 50×50 mm with various thicknesses (2–4–6 mm) to measure RL in the frequency band of 8.8–12 GHz.

4. Experimental setup for measuring reflection loss (RL)

The metal plate is put to reflect the transmitted power of the EM waves as shown in Figure 2. As a consequence, the transmitted power of the EM waves (p_t) is negligible in microwave absorption. Microwave absorption properties of the prepared samples are estimated with the free-space technique as shown in Figure 3. EM waves are generated by a microwave generator in the frequency band of 8.8–12 GHz, where a microwave generator is connected by a WR90 waveguide instrument (IEC Standard R100, X Band). The incident EM waves (p_{in}) are measured by the horn antenna connected to an oscilloscope (Fig. 3), then a prepared

sample is placed on the metal plate at an angle of 45° to measure the reflected power of the EM waves (P_{ref}) by an oscilloscope (Fig. 4). As a result, one can calculate the RL by applying the equation (1) [5, 6]:

$$RL(dB) = 10 \log \frac{P_{in}}{P_{ref}} . \quad (1)$$



a — the metal salts aqueous solution in the flask; *b* — blending totally the whole mixture;
c — heating the solution at 90 °C; *d* — formation of a viscous liquid;
e — formation of a fluffy carbonaceous pyrolyzed mass; *f* — obtaining nanoparticles ferrite

Figure 1. Synthesis of NiZn ferrite nanoparticles by a self-combustion method

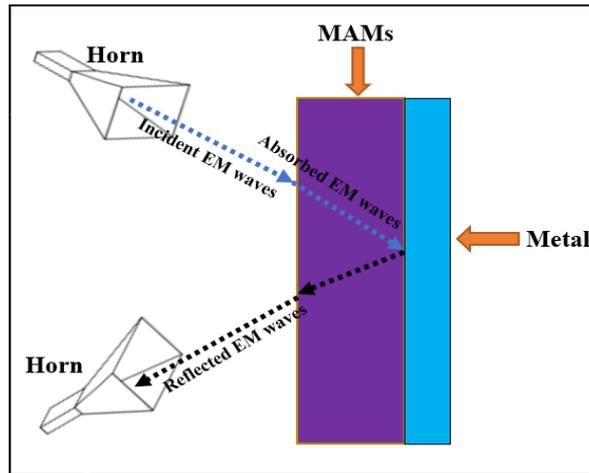
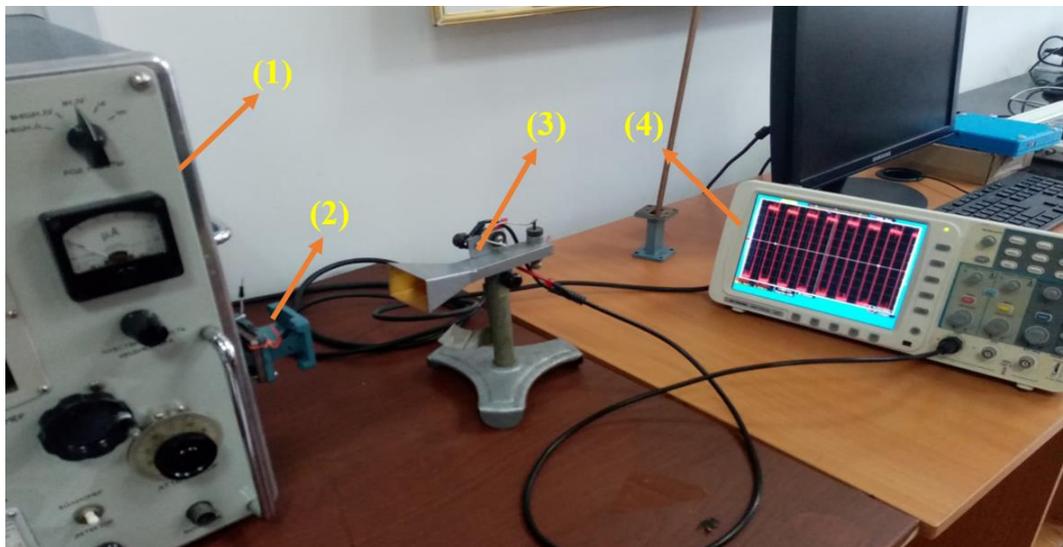


Figure 2. Sketch of the microwave absorption model used to measure reflection loss



1 — microwave generator; 2 — waveguide instrument (IEC Standard R100, X Band);
3 — horn antenna; 4 — oscilloscope

Figure 3. Experimental setup for measuring the incident power of the EM waves by the free-space technique



Figure 4. Experimental setup for measuring the reflected power of the EM waves by the free-space technique

5. Statistical processing of experimental data

Four samples of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4/\text{CI}/\text{CB}$ composites were designed for each thickness (2–4–6 mm) to measure the microwave absorption properties. Statistical processing is used to describe the experimental data. For an estimated statistic, we can use a confidence interval. The purpose of a confidence interval is to supplement the value estimate of the sample within the formation of the uncertainty in this estimate. The method estimate \pm margin of error is used to get an interval based on each sample. The margin of error is given by the equation (2) [7]:

$$\text{Margin of error} = Z^* \cdot \frac{\text{standard deviation}}{\sqrt{n}}, \quad (2)$$

where Z^* is the confidence level which is selected at 95 %. The value of Z^* for a specific confidence level is 1.96; n is the size of the sample that is used to compute the margin of error.

The confidence interval is given by the equation (3) [7]:

$$\text{Confidence interval} = \text{Sample mean} \pm Z^* \cdot \frac{\text{standard deviation}}{\sqrt{n}}. \quad (3)$$

Results and discussion

XRD patterns

The XRD patterns of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$, CI and CB powders are shown in Figure 5. For the $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ pattern, nine diffraction peaks were observed, which conformed to (hkl) planes of (111), (220), (311), (222), (400), (422), (511), (440) and (533), respectively. The ideal spinel structure was observed by the peaks of NiZn ferrite [8]. All the observed peaks of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ matched with the standard XRD pattern (JCPDS, PDF no. 08–0234). On the other hand, for the carbonyl iron pattern, three characteristic peaks were observed, which conformed to (hkl) planes of (100), (200), and (211), respectively. The XRD pattern of carbonyl iron resembles crystallites in which the sample mainly contains α -Fe phase [9]. All the observed peaks of CI matched with the standard XRD pattern (JCPDS, PDF no. 06-0696). Finally, for the CB pattern, two characteristic peaks were observed, which conformed to (hkl) planes of (002) and (100), respectively [10].

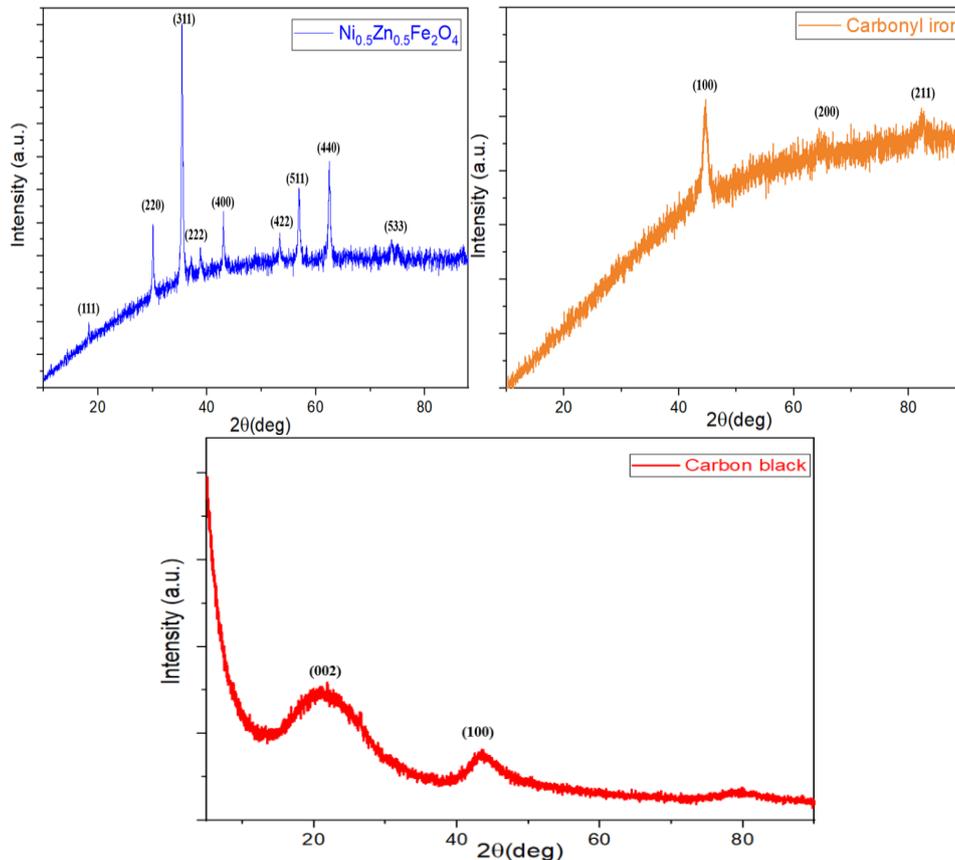


Figure 5. XRD patterns of $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$, carbonyl iron and carbon black

FTIR spectra

The FTIR spectrum of the $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$, CI and CB powders is shown in Figure 6. For the $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$ nanoparticles, two peaks at 565.4 cm^{-1} and 432.3 cm^{-1} are referred to the stretching vibration of (Fe-O), which emphasizes the forming of the metal-oxygen in ferrite-based [11]. On the other hand, the peak at 1630.4 cm^{-1} in $\text{Ni}_{0.5}\text{Zn}_{0.5}\text{Fe}_2\text{O}_4$, CI, and CB is referred to C=O stretching vibration, and the peaks at 2348 cm^{-1} and 3452 cm^{-1} are referred to O-H stretching vibration [12, 13].

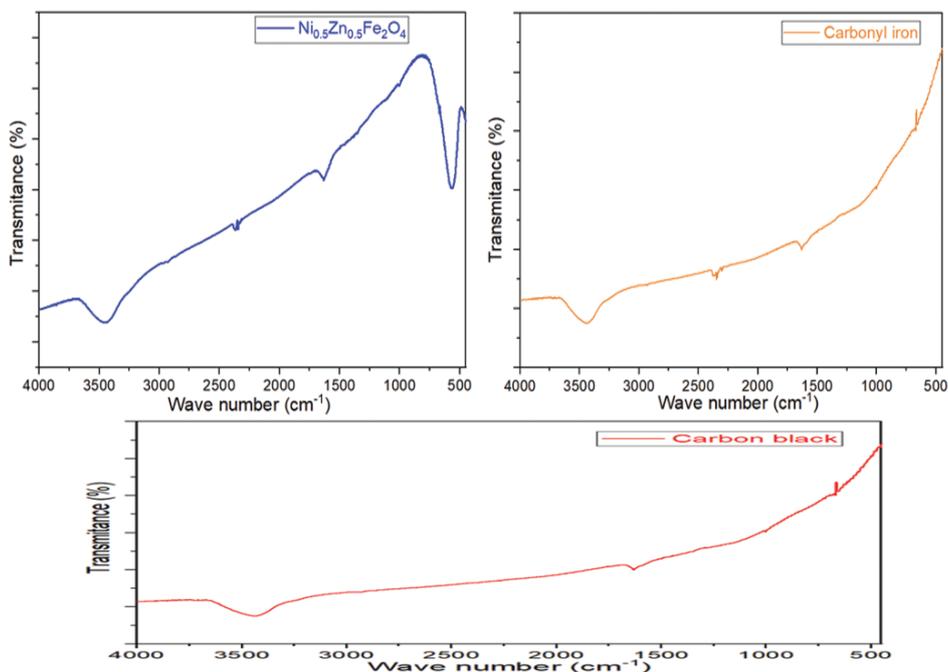
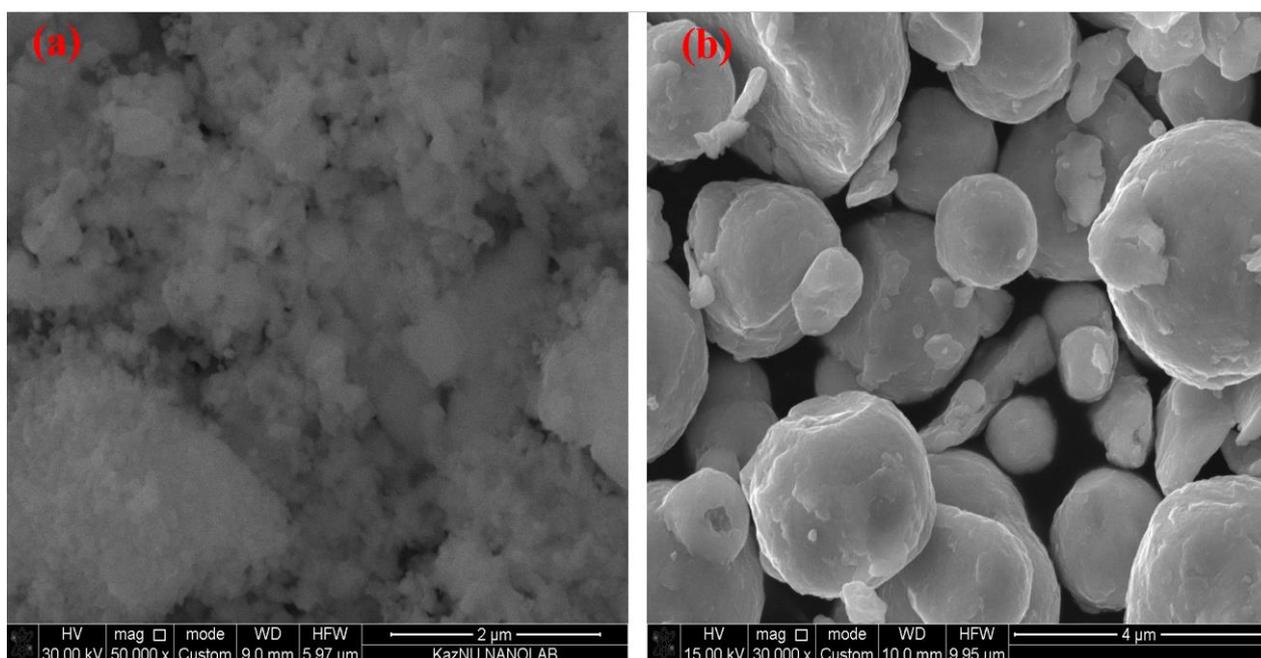


Figure 6. FTIR spectra of NiZn ferrite, carbonyl iron and carbon black

SEM analysis

The powders morphology was investigated by SEM. The agglomerated spherical particles of NiZn ferrite and the spherical particles of carbonyl iron (Fig. 7a, b) were observed with the average diameters to be ranging between 18–52 nm and 0.2–2.4 μm , respectively. On the other hand, the average particle size of carbon black powder (Fig. 7c) was found to be ranging between 75–481 nm.



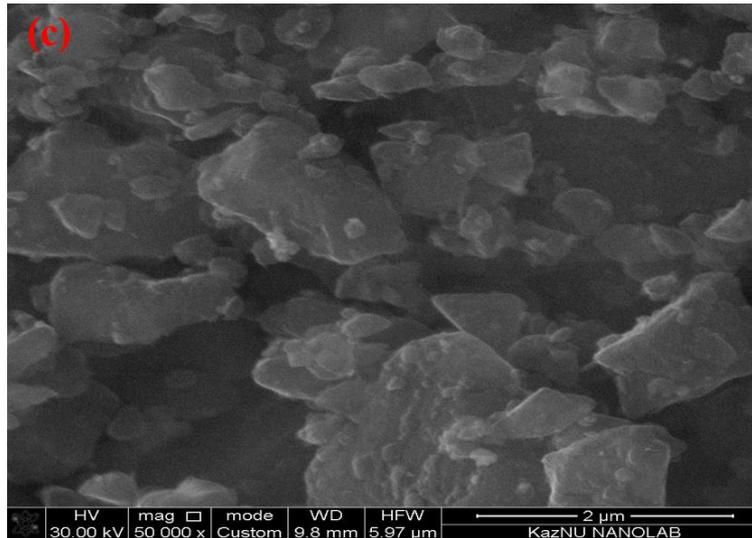


Figure 7. SEM images of (a) NiZn ferrite, (b) carbonyl iron and (c) carbon black

Microwave absorption properties

MA properties of the F/CI/CB composites with various thicknesses (2–4–6 mm) at the weight percentage of the absorber within a paraffin matrix (40 % w/w) were studied. The results of this investigation are exhibited in Figure 8 which illustrates the changing of the RL as a function of the EM wave frequency. The RL of the prepared samples was measured by the variation between the power of incident (p_{in}) and reflected (p_{ref}) EM waves as shown in Figure 3 and Figure 4. The RL curves of the samples were obtained from equation (1). The $BW_{-10\text{ dB}}$, RL_{min} and f_m were deduced from RL curves. Figure 8 shows that the RL attenuation peaks of samples moved to lower frequencies with increasing sample thickness. This phenomenon may be defined by the quarter-wavelength ($\lambda/4$) cancellation model, as shown in equation (4) [14–16]:

$$t_m = \frac{c}{4f_m \sqrt{|\mu_r| |\epsilon_r|}}, \tag{4}$$

where $|\epsilon_r|$ and $|\mu_r|$ are the modulus of the measured complex relative permittivity (ϵ_r) and permeability (μ_r) at matching frequency (f_m), respectively; c is the velocity of light.

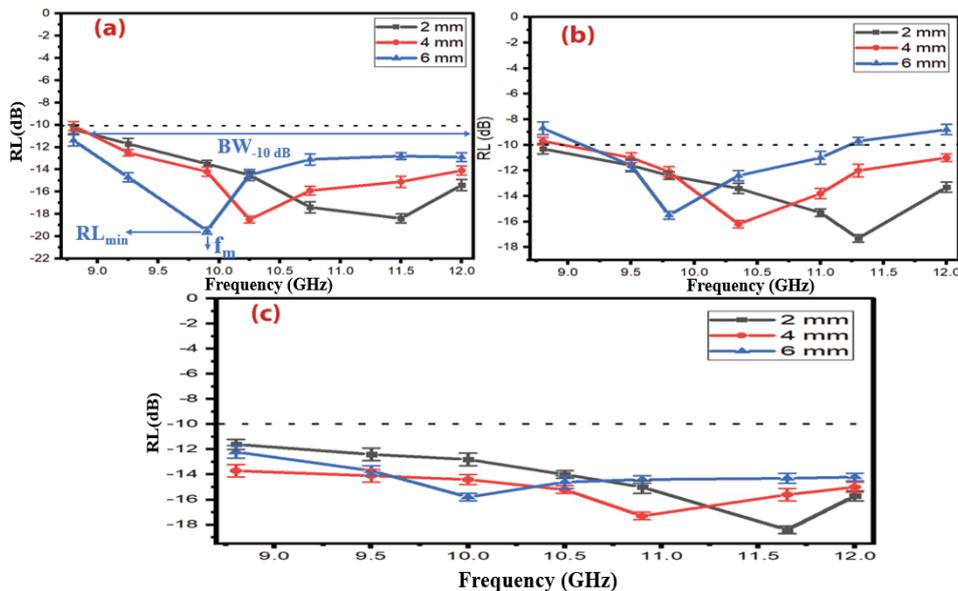


Figure 8. RL curves of (a) F/CI/CB-111 composite; (b) F/CI/CB-112 composite and (c) F/CI/CB-211 composite at various thicknesses (2–4–6 mm)

It can be noticed from equation (4) that the f_m is inversely proportionate to the thickness of an absorber. Furthermore, Table 1 shows the F/CI/CB composites have reasonable surface density, ranging from 3.625 to 4.041 kg/m², and wide bandwidth extending from 2.81 to 3.20 GHz. The absorber of 3.20 GHz bandwidth has a minimal reflection loss of -19.42 dB at the matching frequency of 9.92 GHz with a thickness of 6 mm. One can conclude that the optimal absorption can be accomplished by modifying the absorber thickness of the absorber within a paraffin matrix. In addition to that, one can notice the impact of incorporating Ni_{0.5}Zn_{0.5}Fe₂O₄ and CI (magnetic loss materials) and CB (dielectric loss material) on the MA properties of the prepared absorber. This incorporation leads to an effective and low thickness absorber with a wide bandwidth under -10 dB [17]. Statistical processing was used to describe the experimental data, as the experiments showed the effect of sample density on microwave absorption properties (Table 1), as a result, the confidence interval on the microwave absorption properties was studied.

Table 1

MA behavior of Ni_{0.5}Zn_{0.5}Fe₂O₄/CI/CB composites at various thicknesses (2–6 mm)

Composite samples	t (mm)	RL_{\min} (dB)	f_m (GHz)	$BW_{-10\text{dB}}$ (GHz)	SD (kg/m ²)
F/CI/CB-111	2	-18.32±0.86	11.52±0.25	3.20±0.01	3.861±0.007
	4	-18.54±0.71	10.34±0.36	3.20±0.01	3.874±0.006
	6	-19.42±0.83	9.92±0.14	3.20±0.01	3.892±0.004
F/CI/CB-112	2	-17.36±0.62	11.34±0.37	3.20±0.01	3.625±0.005
	4	-16.33±0.86	10.41±0.16	2.92±0.05	3.643±0.005
	6	-15.58±0.73	9.82±0.23	2.81±0.03	3.652±0.008
F/CI/CB-211	2	-18.46±0.71	11.73±0.13	3.20±0.00	4.012±0.009
	4	-17.31±0.79	10.96±0.38	3.20±0.00	4.027±0.007
	6	-15.82±0.61	10.00±0.35	3.20±0.00	4.041±0.006

Conclusion

F/CI/CB microwave absorbers were synthesized within a paraffin matrix successfully. Ni_{0.5}Zn_{0.5}Fe₂O₄ and CI were used to enhance the mechanism of magnetic loss, while CB was introduced to enhance the mechanism of dielectric loss. As a result, one can notice the impact of combining Ni_{0.5}Zn_{0.5}Fe₂O₄, CI and CB on the MA properties of the absorber. This combination leads to an effective and low thickness absorber with a wide BW_{-10dB}. The results refer that by sufficient control of the thickness of the absorption material and the weight ratio of F/CI/CB, we can design a wideband absorber based on F/CI/CB in the frequency band of 8.8–12 GHz.

References

- Ting, T.H., Yu, R.P., & Jau, Y.N. (2011). Synthesis and microwave absorption characteristics of polyaniline/NiZn ferrite composites in 2–40GHz. *Materials Chemistry and Physics*, 126(1–2), 364–368. <https://doi.org/10.1016/j.matchemphys.2010.11.011>
- Houbi, A., Aldashevich, Z.A., Atassi, Y., Bagasharova Telmanovna, Z., Saule, M., & Kubanych, K. (2021). Microwave absorbing properties of ferrites and their composites: A Review. *Journal of Magnetism and Magnetic Materials*, 529, 167839. <https://doi.org/10.1016/j.jmmm.2021.167839>
- Feng, W., Cao, Y., Gang, J., & Su, W. (2018). Preparation and microwave absorption property of bafe12-txixO19/carbonyl iron powder nanocomposites. *Integrated Ferroelectrics*, 190(1), 63–70. <https://doi.org/10.1080/10584587.2018.1456159>
- Ren, X., Fan, H., & Cheng, Y. (2016). Microwave absorption properties of double-layer absorber based on carbonyl iron/barium hexaferrite composites. *Applied Physics A*, 122(5). <https://doi.org/10.1007/s00339-016-0041-8>
- Bayat, M., Yang, H., Ko, F.K., Michelson, D., & Mei, A. (2014). Electromagnetic interference shielding effectiveness of hybrid multifunctional Fe₃O₄/carbon nanofiber composite. *Polymer*, 55(3), 936–943. <https://doi.org/10.1016/j.polymer.2013.12.042>
- Hong, Y.K., Lee, C.Y., Jeong, C.K., Lee, D.E., Kim, K., & Joo, J. (2003). Method and apparatus to measure electromagnetic interference shielding efficiency and its shielding characteristics in broadband frequency ranges. *Review of Scientific Instruments*, 74(2), 1098–1102. <https://doi.org/10.1063/1.1532540>
- Simundic, A.-M. (2008). Confidence interval. *Biochemia Medica*, 154–161. <https://doi.org/10.11613/bm.2008.015>
- El Nahrawy, A.M., Salah El-Deen, H., Soliman, A.A., & Mosa, W.M.M. (2018). Crystallographic and magnetic properties of Al³⁺Co-doped NiZnFe₂O₄ nano-particles prepared by sol-gel process. *Egyptian Journal of Chemistry*. <https://doi.org/10.21608/ejchem.2018.4504.1397>

- 9 Bahri-Laleh, N., Didehban, K., Yarahmadi, E., Mirmohammadi, S.A., & Wang, G. (2017). Microwave absorption properties of polyaniline/carbonyl iron composites. *Silicon*, 10(4), 1337–1343. <https://doi.org/10.1007/s12633-017-9609-y>
- 10 Hu, E., Hu, X., Liu, T., Fang, L., Dearn, K.D., & Xu, H. (2013). The role of soot particles in the tribological behavior of engine lubricating oils. *Wear*, 304(1-2), 152–161. <https://doi.org/10.1016/j.wear.2013.05.002>
- 11 Kondawar, S.B., & Nandapure, A.I. (2014). Magnetic and electrical properties of zinc-substituted nickel ferrite reinforced conducting polyaniline nanocomposites. *Journal of the Chinese Advanced Materials Society*, 2(3), 186–198. <https://doi.org/10.1080/22243682.2014.934919>
- 12 Sarker, B.K., Kang, N., & Khondaker, S.I. (2014). High performance semiconducting enriched carbon nanotube thin film transistors using metallic carbon nanotubes as electrodes. *Nanoscale*, 6(9), 4896. <https://doi.org/10.1039/c3nr06470k>
- 13 Kim, S.Y., Kwon, S.H., Liu, Y.D., Lee, J.-S., You, C.-Y., & Choi, H.J. (2013). Core-shell-structured cross-linked poly(glycidyl methacrylate)-coated carbonyl iron microspheres and their magnetorheology. *Journal of Materials Science*, 49(3), 1345–1352. <https://doi.org/10.1007/s10853-013-7818-3>
- 14 Wang, S., Jiao, Q., Shi, Q., Zhu, H., Feng, T., Lu, Q., Feng, C., Li, H., Shi, D., & Zhao, Y. (2020). Synthesis of porous nitrogen-doped graphene decorated by γ -Fe₂O₃ nanorings for enhancing microwave absorbing performance. *Ceramics International*, 46(1), 1002–1010. <https://doi.org/10.1016/j.ceramint.2019.09.064>
- 15 Shu, R., Zhang, J., Guo, C., Wu, Y., Wan, Z., Shi, J., Liu, Y., & Zheng, M. (2020). Facile synthesis of nitrogen-doped reduced graphene oxide/nickel-zinc ferrite composites as high-performance microwave absorbers in the X-band. *Chemical Engineering Journal*, 384, 123266. <https://doi.org/10.1016/j.cej.2019.123266>
- 16 Jaiswal, R., Agarwal, K., Kumar, R., Kumar, R., Mukhopadhyay, K., & Prasad, N.E. (2020). EMI and microwave absorbing efficiency of polyaniline-functionalized reduced graphene oxide/ γ -Fe₂O₃/epoxy nanocomposite. *Soft Matter*, 16(28), 6643–6653. <https://doi.org/10.1039/d0sm00266f>
- 17 Ali, N.N., Atassi, Y., Salloum, A., Malki, A., Jafarian, M., & Almarjeh, R.K. (2019). Lightweight broadband microwave absorbers of core-shell (polypyrrole/nickel-zinc ferrite) nanocomposites in the X-band: Insights on Interfacial Polarization. *Journal of Materials Science: Materials in Electronics*, 30(7), 6876–6887. <https://doi.org/10.1007/s10854-019-01002-y>

А. Хуби, А.А. Жарменов, И. Атасси,
Ж.Т. Багашарова, С. Мырзалиева, Б.А. Кәрібаев

Үштік композиттердің синтезі және микротолқынды сіңіру қасиеттері (Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB)

Жұмыста феррит NiZn/темір карбонил/ қара көміртек (Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB) үштік композиттері екі кезеңде алынды: біріншіден, Ni_{0,5}Zn_{0,5}Fe₂O₄ өздігінен жану арқылы алынды. Одан кейін ұнтақтау шарлары арқылы CB, CI және Ni_{0,5}Zn_{0,5}Fe₂O₄ араластыруымен операция жалғасты. Әр түрлі қалыңдықтағы (2–4–6 мм) Ni_{0,5}Zn_{0,5}Fe₂O₄ (1:1:1, 1:1:2 және 2:1:1) үш түрлі салмақ қатынасы дайындалды. Абсорберлер салмағы бойынша 40 % парафинді балауыз матрицасында салмақ қатынасы бар композиттерді (Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB) дисперсиялау арқылы жасалды. Үлгілерді сипаттау үшін рентгендік дифрактометрия және FTIR спектроскопиясы қолданылады. Ұнтақтардың морфологиясы SEM арқылы зерттелді. Функционалдық сипаттама 8,8–12 ГГц жиілік диапазонында микротолқынды сіңіру қасиеттерін өлшеу арқылы орындалды. Микротолқынды сіңіру материалдары 2,81–3,20 ГГц диапазонында –10 дБ-ден төмен кең өткізу қабілеттілігін және 3,625–4,041 кг/м² диапазонында қолайлы бет тығыздығын көрсетті. Өткізу жолағы 3,20 ГГц, қалыңдығы 6 мм болатын абсорбер 9,92 ГГц сәйкес жиілікте –19,4 дБ шағылудың минималды жоғалуына ие.

Кілт сөздер: NiZn ферриті, карбонилді темір, қара күйе, шағылысу жоғалтуы, сіңіру жолағы, сәйкестік жиілігі, парафинді балауыз матрицасы, беттік тығыздығы.

А. Хуби, А.А. Жарменов, И. Атасси,
Ж.Т. Багашарова, С. Мырзалиева, Б.А. Кәрібаев

Синтез и микроволновые поглощающие свойства тройных композитов (Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB)

В настоящей работе были получены тройные композиты феррит NiZn/карбонил железа/сажа (Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB) в два этапа. Сначала методом самовозгорания был получен Ni_{0,5}Zn_{0,5}Fe₂O₄. После этого операция продолжалась путем перемешивания CB, CI и Ni_{0,5}Zn_{0,5}Fe₂O₄ через мелющие шары. Были изготовлены три различных весовых соотношения Ni_{0,5}Zn_{0,5}Fe₂O₄/CI/CB (1:1:1, 1:1:2 и 2:1:1) различной толщины (2–4–6 мм). Поглотители изготавливали путем диспергирования композитов

(Ni_{0,5}Zn_{0,5}Fe₂O₄/Cl/CB) с массовым соотношением в матрице парафинового воска 40 % по массе. Рентгеновская дифрактометрия и FTIR-спектроскопия использовались для характеристики образцов. Морфология порошков была исследована с помощью СЭМ. Функциональная характеристика получена путем измерения характеристик поглощения микроволн в диапазоне частот 8,8–12 ГГц. Материалы, поглощающие микроволновое излучение (ПМИ), показывали широкую полосу пропускания ниже –10 дБ в диапазоне 2,81–3,20 ГГц и приемлемую поверхностную плотность в пределе 3,625–4,041 кг/м². Поглотитель с полосой пропускания 3,20 ГГц имеет минимальные потери на отражение — 19,4 дБ на частоте согласования 9,92 ГГц при толщине 6 мм.

Ключевые слова: феррит NiZn, карбонильное железо, сажа, потери на отражение, ширина полосы поглощения, частота согласования, парафиновая матрица, поверхностная плотность.

Information about authors*

Houbi, Anas (*corresponding author*) — 3rd year PhD student, Department of Chemical Technology of Inorganic Substances, Al-Farabi Kazakh National University, Al-Farabi street, 71, 050040, Almaty, Kazakhstan; e-mail: anashoubi@gmail.com; <https://orcid.org/0000-0001-9282-774X>;

Zharmenov, Abdurassul Aldashevich — General Director of the RSE “National Center on Complex Processing of Mineral Raw Materials of the Republic of Kazakhstan”, Dr. Sci. (Eng.), Prof., academician of the NAS of RK, State Premium Double Laureate, Jandossov street, 67, 050036, Almaty, Kazakhstan; e-mail: nc@cmrp.kz; <https://orcid.org/0000-0001-5651-5343>;

Atassi, Yomen — Professor of Chemistry of Materials, Director of Materials Science Laboratory, Higher Institute for Applied Science and Technology (HIASST), Barzeh street, Damascus, Syria; e-mail: yomen.atassi@gmail.com; <https://orcid.org/0000-0002-1338-4993>;

Bagasharova, Zhenisgul Telmanovna — Candidate of Technical Sciences, RSE “National center for complex processing of mineral raw materials of the Republic of Kazakhstan”, senior lecturer at Al-Farabi Kazakh National University, Al-Farabi street, 71, 050040, Almaty, Kazakhstan; e-mail: zh.t_bagasharova@mail.ru; <https://orcid.org/0000-0001-8996-8656>;

Mirzalieva, Saule — Doctor of chemical Sciences, Professor, Head of the Department for training of scientific personnel national center for complex processing of mineral raw materials of the Republic of Kazakhstan, professor at Al-Farabi Kazakh National University, Al-Farabi street, 71, 050040, Almaty, Kazakhstan; e-mail: saulekerchaiz@mail.ru; <https://orcid.org/0000-0003-2997-0716>;

Karibayev, Beibit Abdirbekovich — Acting assistant professor of Physics and Technology, Al-Farabi Kazakh National University, Al-Farabi street, 71, 050040, Almaty, Kazakhstan; e-mail: beibitkaribaev7@gmail.com; <https://orcid.org/0000-0003-1057-0296>

*The author's name is presented in the order: *Last Name, First and Middle Names*